

Govt. College, Alewa (Jind)

Session: 2024-2025 (Even Semester)

Lesson Plan

Name of the Teacher: Dr. MANJEET SINGH, ASSISTANT PROFESSOR

Subject: PHYSICS (Theory) Class: B. Sc. 6th Sem.

Physics Paper: I (Solid State and Nano Physics)

Paper's Code: NPH06(I)

Max. Marks: 50 (External Assessment: 40 Marks, Internal Assessment: 10 Marks)

Contact Hours Per Week: 2

Physics Paper: II (Atomic and Molecular Spectroscopy)

Paper's Code: NPH06(II)

Max. Marks: 50 (External Assessment: 40 Marks, Internal Assessment: 10 Marks)

Contact Hours Per Week: 2

Sr. No.	Months	Dates	Topic
1	January 2025	15-01-2025 to 31-01-2025	Crystalline and glassy forms, liquid crystals, crystal structure, periodicity, lattice and basis, crystal translational vectors and axes. Unit cell and Primitive Cell, Wigner Seitz primitive Cell, symmetry operations for a two dimensional crystal, Bravais lattices in two and three dimensions. Crystal planes and Miller indices, Interplaner spacing, Crystal structures of Zinc Sulphide, Sodium Chloride and Diamond, X-ray diffraction, Bragg's Law, experimental X-ray diffraction methods. K-space and reciprocal lattice and its physical significance, reciprocal lattice vectors, reciprocal lattice to a simple cubic lattice, b.c.c. and f.c.c.
2	February 2025	01-02-2025 to 28-02-2025	Historical introduction, Survey of superconductivity, Super conducting systems, High Tc Super conductors, Isotopic Effect, Critical Magnetic Field, Meissner Effect, London Theory and Pippards' equation, Classification of Superconductors (type I and Type II), BCS Theory of Superconductivity, Flux quantization, Josephson Effect (AC and DC), Practical Applications of superconductivity and their limitations, power application of superconductors. Definition, Length scale, Importance of Nano-scale and technology, History of Nanotechnology Benefits and challenges in molecular manufacturing. Molecular assembler conce Understanding advanced capabilities. Vision and objective of Nano-technology, Nanotechnology in different field, Automobile, Electronics, Nano-biotechnology, Materials, Medicine.
3	March 2025	01-03-2025 to 31-03-2025	Introduction of early observations, emission and absorption spectra, atomic spectra, wave number, spectrum of Hydrogen atom in Balmer series, Bohr atomic model (Bohr's postulates), spectra of Hydrogen atom, explanation of spectral series in Hydrogen atom, un-quantized states and continuous spectra, spectral series in absorption spectra effect of nuclear motion on line spectra (correction of finite nuclear mass), variation in Rydberg constant due to finite mass, short comings of Bohr's theory, Wilson sommerfeld quantization rule, de-Broglie interpretation of Bohr quantization law, Bohr's corresponding principle, Sommerfeld's extension of Bohr's model, Sommerfeld relativistic correction, Short comings of Bohr-Sommerfeld theory, Vector atom model; space quantization, electron spin, coupling of orbital and spin angular

			momentum, spectroscopic terms and their notation, quantum numbers associated with vector atom model, transition probability and selection rules. Orbital magnetic dipole moment (Bohr magneton), behavior of magnetic dipole in external magnetic field; Larmors' precession and theorem. Penetrating and Non-penetrating orbits, Penetrating orbits on the classical model; Quantum defect, spin orbit interaction energy of the single valence electron, spin orbit interaction for penetrating and non-penetrating orbits. quantum mechanical relativity correction, Hydrogen fine spectra,
4	April 2025	01-04-2025 to 30-04-2025	Main features of Alkali Spectra and their theoretical interpretation, term series and limits, Rydberg-Ritz combination principle. Absorption spectra of Alkali atoms, observed doublet fine structure in the spectra of alkali metals and its Interpretation, Intensity rules for doublets, comparison of Alkali spectra and Hydrogen spectrum. Essential features of spectra of Alkaline-earth elements, Vector model for two valence electron atom: application of spectra, Coupling Schemes; LS or Russell – Saunders Coupling Scheme and JJ coupling scheme, Interaction energy in L-S coupling (sp, pd configuration), Lande interval rule, Pauli principal and periodic classification of the elements, Interaction energy in JJ Coupling(sp, pd configuration), equivalent and non-equivalent electrons,
5	May 2025	01-05-2025 Onwards	Two valence electron system-spectral terms of non-equivalent and equivalent electrons, comparison of spectral terms in L-S And J-J coupling. Hyperfine structure of spectral lines and its origin; isotope effect, nuclear spin. Paschen-Back effect of a single valence electron system, Weak field Stark effect of Hydrogen atom. General Considerations, Electronic States of Diatomic Molecules, Rotational Spectra (Far IR and Microwave Region), Vibrational Spectra (IR Region), Rotator Model of Diatomic Molecule, Raman Effect, Electronic Spectra. Revision Class Test

Dr. Manjeet Singh
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